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Extraction and Identification of Zoochemicals in Marine Sponge *Hyattella intestinalis* (Lamarck, 1814) (Phylum: Porifera) using GC-MS Technique

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Abstract: Animals provide secondary metabolites with natural bioactive properties, similar to plant secondary metabolites. Most primitive multicellular animals like sponges synthesize secondary metabolites like alkaloids, terpene etc. to be used in ecological aspects. The aim of the present study was to extract and identify Zoochemicals from *Hyattella intestinalis* (Lamarck, 1814), a marine sponge, by using GC-MS techniques. 70% hydroethanolic extract of whole portion of *Hyattella intestinalis* was prepared by using the Soxhlet apparatus with a 3 h duration at 50 to 60°C. At the end of the experiment, a light greenish yellow colour of Zoo-extract was obtained and subjected to qualitatively identified for alkaloids and terpenoids. Sixty (60) compounds were identified by GC-MS analysis. Some of the identified compounds of *Hyattella intestinalis* have naturally reducing agents and biological active properties while using nanoparticle synthesis to reduce heavy metal pollutants in environmental applications. As the *Hyattella intestinalis* involves Zoochemicals, safety can be addressed easily with environmental impacts.

Keywords: *Hyattella intestinalis*, Zoochemicals, GC-MS analysis, Zoo-extraction, Bioactive properties

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Introduction

There is a growing interest in utilising marine natural product scaffolds' richness and complexity for rational drug development. Marine-based pharmaceuticals have the potential to identify novel entities that may help in the treatment of biological activities (Khalifa *et al.*, 2019). Marine

sponges contributed nearly 30 per cent of all natural products, and the initial discoveries from marine sponges led to the belief that it would not be long before true marine-derived drugs were developed; for instance, the discovery and identification of spongothymidine and spongo-

uridine from the sponge of *Tethya crypta*, which are used for antiviral activity, were attributed to marine sponges (Newman *et al.*, 2000). The study of marine sponges and bioactive compounds obtained from them is a rapidly increasing area of zoo-chemicals derived from marine organisms (Amudha *et al.*, 2017; Muthiyan *et al.*, 2020).

From marine algae, sponges, and sponge-associate microorganisms, a number of useful chemical compounds with a wealth of potential bioactivities i.e. medicines, have been identified and shown to have therapeutic utility (Mille *et al.*, 2010). Over the last 50 years, natural marine products of sponges have drawn the attention of chemists and biologists across the globe as a source of possible medications (Yalcin, 2007; Mohanasundaram *et al.*, 2021). Using GC-MS methods, the present study aimed to extract and identify zoochemicals from *Hyattella intestinalis*.

Materials and Methods

Collection of sponge:

The sponge was collected by hand-picking at the east coast of Mallipattinam village, Thanjavur district, Tamil Nadu, India. The collected sponge was identified by existing literature (Cook and Bergquist, 2002; Srinivasan Balakrishnan, 2007; Sivaleela, 2014; Varsha *et al.*, 2020).

Extraction of *Hyattella intestinalis* and GC-MS analysis:

Hyattella intestinalis was separately cut into pieces and 70% hydroalcoholic extract was prepared by using Soxhlet extraction (3 h and 50 to 60°C) The extract was subjected to zoochemical analysis (Harborne, 1973) and bioactive compounds were identified using GC-MS technique. GC MS analysis was carried out on Shimadzu 2010 plus comprising a AOC-20i auto sampler and gas chromatograph interfaced to a mass spectrometer instrument. Interpretation on GC-MS was conducted using the database of National Institute Standard and Technology (NIST).

Results and Discussion

Zoochemical extraction technique and preliminary

zoo chemical identified:

Marine sponges are preliminary invertebrates that can be found in temperate, polar and tropical regions. They are major contributors of bioactive compounds (Varijakzhan *et al.*, 2021). In the present study screening of Zoochemicals was performed from *Hyattella intestinalis* using extraction and GC-MS techniques. Zoochemicals are animal equivalent of phytochemicals in plants (<http://lpi.oregonstate.edu/infocenter/phytochemicals.html>). The Soxhlet method involves boiling the *Hyattella intestinalis* whole portion with 70% ethanol, with a duration of 3 h and temperature from 50 to 60°C. *Hyattella intestinalis* 20 g sample extract with 200 ml and 70% ethanol after 3 h produced light greenish yellow colour extract which was concentrated to 20 ml (Fig. 1) and further qualified for alkaloids, terpenoids (Table 1; Fig. 2) and GC-MS analysis. Abubakar and Haque (2020) extracted sponge compound which are involved in secondary metabolites (Mohanasundaram *et al.*, 2021). Muthiyan *et al.* (2020) reported the major zoo-chemical constituent of the sponge present as alkaloids. The present study derives support from observations of Gupta (2019) who has extracted terpenes and alkaloid as bioactive secondary metabolites from the genus of *Haliclona*.

GC-MS analysis of *Hyattella intestinalis* Zooextract:

The present study evaluated the presence of 60 Zoochemicals from 70% ethanolic (hydro-alcoholic) extract of *Hyattella intestinalis* by using GC-MS techniques (Table 2; Fig. 2). Interpretation on GC-MS was conducted using the database of National Institute Standard and Technology (NIST). Table 2 illustrates the active principles with their RT (Retention time), name of the compound, molecular formulae, molecular weight (g/mol) by comparing a query mass spectrum with reference data in a library of spectrum matching (Wei *et al.*, 2014).

The potential of marine sponges producing bioactive compounds having biological active properties has been reported by Perdicaris *et al.* (2013). Marine sponges are considered as rich



Fig. 1: Extraction of Zoochemicals using Soxhlet method and qualified.

Table 1: Preliminary qualified in Zoochemicals of *Hyattella intestinalis*

| Zoochemicals | Observation | Result (figure 2) |
|--------------|---|---------------------------|
| Alkaloids | White precipitate | Higher concentration (++) |
| Terpenoids | Appearance of two layer in reddish brown colour | Presence (+) |

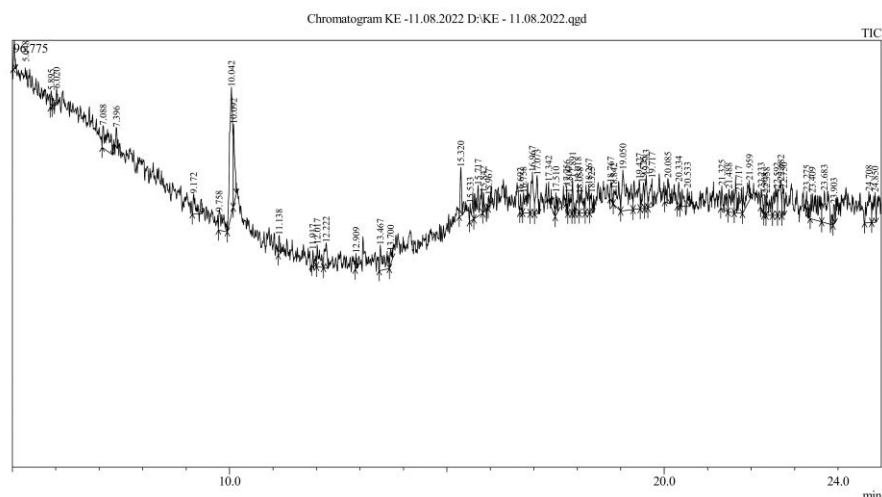


Table 2: Identification of Zoochemicals using GC-MS techniques in *Hyattella intestinalis* Zooextract

| Peak # | R. Time | A/H ratio | Molecular weight (g/mol) | Molecular formula | Molecular name |
|--------|---------|-----------|--------------------------|---|--|
| 1 | 5.048 | 1.73 | 99 | C ₆ H ₁₃ N | 1,2,2-Trimethylcyclopropylamine |
| 2 | 5.895 | 1.63 | 262 | C ₁₇ H ₂₆ O ₂ | [1,1'-Bibicyclo[2.2.2]Octane]-4-carboxylic acid |
| 3 | 6.020 | 2.14 | 41 | C ₂ H ₃ N | Acetonitrile |
| 4 | 7.088 | 8.79 | 42 | CH ₃ BO | Borane carbonyl |
| 5 | 7.396 | 1.62 | 40 | Ar | Argon |
| 6 | 9.172 | 4.75 | 208 | C ₁₃ H ₂₀ O ₂ | 3-Buten-2-one, 4-(2,2,6-Trimethyl-7-oxabicyclo[4.1.0]hept-1-yl)- |
| 7 | 9.758 | 7.02 | 85 | C ₃ H ₃ NO ₂ | Acetic acid, cyano- |
| 8 | 10.042 | 4.94 | 222 | C ₁₂ H ₁₄ O ₄ | Diethyl Phthalate |
| 9 | 10.092 | 2.11 | 339 | C ₁₇ H ₉ NO ₃ S ₂ | 3-Phenyl-5-phthalidylidenerhodanine |
| 10 | 11.138 | 1.73 | 85 | C ₃ H ₃ NO ₂ | 2-Cyanoacetic acid |
| 11 | 11.917 | 2.35 | 251 | C ₁₆ H ₁₃ NO ₂ | Pyrano[3,4-B]Indol-3(9H)-One, 1-(4-Pentynyl) |
| 12 | 12.017 | 3.98 | 84 | C ₄ H ₄ O ₂ | But-3-Ynoic acid |
| 13 | 12.222 | 3.98 | 206 | C ₇ H ₁₁ BrO ₂ | 1,3-Dioxolane, 2-(3-bromo-3-buten-1-yl)- |
| 14 | 12.909 | 0.99 | 112 | C ₄ H ₄ N ₂ O ₂ | 1H-Imidazole-4-Carboxylic acid |
| 15 | 13.467 | 5.56 | 114 | C ₇ H ₁₄ O | 2-Methyl-5-hexen-3-ol |
| 16 | 13.700 | 2.99 | 104 | C ₄ H ₈ O ₃ | 2-Methoxy-1,3-dioxolane |
| 17 | 15.320 | 2.23 | 458 | C ₂₀ H ₄₂ O ₄ Si ₄ | Silane, [[4-[1,2-Bis[(Trimethylsilyl)Oxy]Ethyl]-1,2-Phenylene]Bis(Oxy)]BIS[Trimethyl |
| 18 | 15.533 | 4.03 | 237 | C ₁₂ H ₂₅ B ₂ NO ₂ | Caprolactone oxime, (NB)-O-[(diethylboryloxy)(ethyl)boryl]- |
| 19 | 15.717 | 5.93 | 361 | C ₂₂ H ₃₅ NO ₃ | Glycine, N-(4-ethylbenzoyl)-, undecyl ester |
| 20 | 15.842 | 1.34 | 224 | C ₃ H ₉ AlCsF | Cesium trimethylfluoro) Aluminate |
| 21 | 15.967 | 1.11 | 206 | C ₁₃ H ₁₈ O ₂ | Ethanone, 1-[4-[(1,1-Dimethylethoxy)Methyl]Phenyl] |
| 22 | 16.692 | 1.46 | 355 | C ₂₂ H ₁₃ NO ₄ | 3-(3-OXO-3H-Benzo[F]Chromen-2-yl)-2,4(1H,3H)-Quinolinedione |
| 23 | 16.758 | 8.23 | 153 | C ₉ H ₁₅ NO | Hexanenitrile, 2-(2-Methoxyethylidene)-, (E) |
| 24 | 16.967 | 3.62 | 458 | C ₂₀ H ₄₂ O ₄ Si ₄ | 4-(3,4-Bis[(Trimethylsilyl)OX |
| 25 | 17.073 | 3.07 | 353 | C ₁₆ H ₂₃ NO ₄ Si | 3-(4-Hydroxy-3-methoxyphenyl)-2-isothiocyanatopropionic acid, ethyl ester, TMS |
| 26 | 17.342 | 3.91 | 222 | C ₆ H ₁₈ O ₃ Si ₃ | Cyclotrisiloxane, Hexamethyl- |
| 27 | 17.510 | 2.08 | 206 | C ₁₀ H ₂₂ S ₂ | t-Butyl n-hexyl disulfide |
| 28 | 17.756 | 0.89 | 153 | C ₉ H ₁₅ NO | 3-Cyano-1-Methoxyhept-2-ENE |
| 29 | 17.800 | 3.76 | 254 | C ₁₅ H ₁₄ N ₂ O ₂ | 4-(Methoxymethyl)-6-Methyl-2-Phenoxynicotinonitrile |
| 30 | 17.891 | 1.60 | 297 | C ₉ H ₉ Cl ₂ NO ₄ S | 2,4-Dichloro-5-Dimethylsulfamoyl-benzoic acid |
| 31 | 18.018 | 2.49 | 131 | C ₄ H ₂ ClNO ₂ | Isoxazole-5-carbonyl chloride |
| 32 | 18.058 | 5.84 | 111 | C ₇ H ₁₃ N | N-(1,1-Dimethyl-2-propynyl)-N,N-dimethylamine |
| 33 | 18.267 | 3.96 | 134 | C ₄ H ₆ O ₅ | Methyltartronic acid |
| 34 | 18.325 | 2.71 | 237 | C ₁₂ H ₂₅ B ₂ NO ₂ | Caprolactimether, (NB)-O-[(Diethylboryloxy)(ETHYL)BORYL] |

| | | | | | |
|----|--------|------|-----|---|--|
| 35 | 18.767 | 2.41 | 234 | C ₅ H ₉ CsO ₂ | Propanoic acid, 2,2-dimethyl-, cesium salt |
| 36 | 18.842 | 9.01 | 268 | C ₁₇ H ₁₆ O ₃ | 2-(4-Benzoylphenyl)-2-Methyl-1,3-Dioxole |
| 37 | 19.050 | 8.10 | 384 | C ₁₂ H ₃₆ O ₄ Si ₅ | Pentasiloxane, Dodecamethyl |
| 38 | 19.427 | 6.22 | 251 | C ₈ H ₇ Cl ₂ NO ₄ | 2-(2,4-Dichloro-6-Nitrophenoxy)Ethanol |
| 39 | 19.525 | 2.98 | 253 | C ₁₁ H ₇ N ₇ O | 2-Propenenitrile, 2-Methyl-3-(9-OXO-9H-[1,2,4]Triazolo[4',3':2,3]Pyr idazino[6,1-C][1,2,4]Triazin-8-YL)- |
| 40 | 19.583 | 2.68 | 223 | C ₁₂ H ₂₁ NOSi | Paredrine tms |
| 41 | 19.717 | 3.62 | 209 | C ₇ H ₁₆ NO ₂ PS | 5,5-Dimethyl-2-(Dimethylamino)-1,3,2-Dioxaphosphorinane 2-Sulphide |
| 42 | 20.085 | 3.71 | 296 | C ₈ H ₂₄ O ₄ Si ₄ | Cyclotetrasiloxane, octamethyl- |
| 43 | 20.334 | 2.13 | 253 | C ₁₁ H ₇ N ₇ O | 9H-[1,2,4]Triazolo[4',3':2,3]Pyridazin |
| 44 | 20.533 | 4.84 | 320 | C ₁₉ H ₂₈ O ₄ | Pentanedioic acid, (2,4-di-t-butylphenyl) mono-ester |
| 45 | 21.325 | 4.49 | 368 | C ₁₃ H ₃₆ O ₄ Si ₄ | 3-Butoxy-1,1,1,5,5,5-hexamethyl-3-(trimethylsiloxy)trisiloxane |
| 46 | 21.488 | 5.65 | 353 | C ₂₀ H ₂₃ N ₃ O ₃ | Carbamic acid, N-[10,11-dihydro-5-(2-methylamino-1-oxoethyl)-3-5H-dibenzo[b,f]azepinyl]-, ethyl ester |
| 47 | 21.717 | 7.46 | 296 | C ₁₅ H ₃₂ Si ₃ | Cyclohexa-1,4-Diene, 1,3,6-Tris(Trimethylsilyl)- |
| 48 | 21.959 | 6.37 | 396 | C ₂₂ H ₁₉ N ₅ O | 1-(4-Acetamidoanilino)-3,7-dimethylbenzo[4,5]imidazo[1,2-a]pyridine-4-carbonitrile |
| 49 | 22.233 | 2.32 | 206 | C ₄ H ₉ F ₃ O ₂ SSi | Methanesulfinic acid, Trifluoro-, trimethylsilyl ester |
| 50 | 22.308 | 2.48 | 253 | C ₁₂ H ₁₅ NO ₅ | Glycine, N-[[[(4-Methoxyphenyl)Methoxy]Carbonyl]-N-Methyl |
| 51 | 22.358 | 4.42 | 291 | C ₂₀ H ₂₁ NO | 1(2H)-Naphthalenone, 2-(3,3-Dimethyl-2-Phenyl-2-Aziridinyl)-3,4-Dihydro |
| 52 | 22.592 | 4.92 | 234 | C ₁₁ H ₂₂ OS ₂ | 1,3-Dithiane-2-Ethanol, .Alpha.-Ethyl-2-(1-Methylethyl) |
| 53 | 22.682 | 4.24 | 460 | C ₂₉ H ₄₈ O ₄ | Phthalic acid, propyl octadecyl ester |
| 54 | 22.750 | 2.58 | 364 | C ₁₈ H ₁₂ F ₃ O ₃ P | Tris(4-fluorophenyl) phosphite |
| 55 | 23.275 | 1.63 | 417 | C ₂₆ H ₄₃ NO ₃ | Glycine, N-(4-ethylbenzoyl)-, pentadecyl ester |
| 56 | 23.409 | 9.70 | 253 | C ₁₃ H ₁₉ NO ₂ S | Benzothiophene-3-carboxylic acid, 4,5,6,7-tetrahydro-2-amino-6-ethyl-, ethyl ester |
| 57 | 23.683 | 7.77 | 282 | C ₁₆ H ₁₄ N ₂ O ₃ | Benzofuro[3,2-B]Pyridine-1(2H)-Carboxylic acid, 3-Cyano-4-Methyl-, Ethyl ester |
| 58 | 23.903 | 1.94 | 280 | C ₉ H ₂₈ O ₂ Si ₄ | Trimethylsilyl-di(trimethylsiloxy)-silane |
| 59 | 24.708 | 5.55 | 234 | C ₅ H ₉ CsO ₂ | Cesium pivalate |
| 60 | 24.850 | 3.59 | 374 | C ₂₄ H ₂₂ O ₄ | Phthalic acid, 3,4-dimethylphenyl 3,5-dimethylphenyl ester |

and supported by studies *in silico*, *in vitro* and *in vivo* studies to be recommended as eco-friendly (Mohanasundaram *et al.*, 2017).

Conclusion

Hyattella intestinalis was positive for alkaloids and

terpenoids while the conformation of 60 Zoochemical compounds was identified using GC-MS techniques. *Hyattella intestinalis* naturally reduces agents, the exhibits stimulating pharmacological activity *in vitro*, *in vivo*, *in silico* studies, and has social relevance because society requires harmless but effective drug/reducing agents that can offer advancements to chemical drugs.

Because *Hyattella intestinalis* involves Zoochemicals, safety can be easily addressed with environmental impacts and scientifically with the results of the current study.

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